

Influence of swelling spruce fibers with different organic and inorganic solvents

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In cooperation with...

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Fractionation

- Sawmill by-products are used as raw material
- Material is pre-treated chemi-mechanically and thermally
- Components are fractionated to be transformed into a new form
- During forming fibers are re-arranged and re-compacted



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What is the purpose...

- Long-term CO₂ storage
- Composites without nondegradable synthetic adhesives
- Biodegradable and environmentally friendly building material
- Improve the value chain of sawmill by-products (wood chips, sawdust, etc.)











Background and aim of the study...

Enhancing the characteristics of **reinforcement material for biocomposites**

- Working with by-products of the wood industry
- Pulping with "green" solvents
- Swelling pulping material to enhance fiber properties
 - Well established method in the paper industry
 - Makes the fibers more accessible and increases the uptake capacity of solvents
 - Important for solvents which can act as an additive carrier for matrix material

Finding sufficient swelling agents to prepare the fibers for impregnation with additives and/or matrix material, to create an **intermediate product for biocomposites**.











Why peracetic acidpulping (PAA)?

- Fibrillation process without severe mechanical treatment
- Initial fiber length almost completely preserved
- Pulping at 80-100°C moderate temperatures
- Selective extraction of lignin
- Preserving hemicellulose to enhance the binding ability^[2-4] – holocellulose pulp



Why swelling?

- Increases surface area for hydrogen bonding, enhances fiber bonding in variable products
- Improves fiber flexibility and elasticity
- Enhances the accessibility and uptake of solvents (with additives or matrix material)
- Improves tensile strength and tear resistance





Feedstock

- Spruce wood chips (softwood)
- Different particle sizes (1mm < x < 5cm)
- Air dried and stored at controlled temperature and humidity











Methods

PAA – process Peracetic Acid^[5]



Holocellulose pulp





Methods

Swelling Different solvents



Methods for analysis

- Liquid retention value (LRV)^[6]
 - Capacity of the fiber for liquid uptake

Microscopy

- Optical analysis of the fibers
- Crystallinity Index
 - Determination of crystalline and amorphous content
 - Wide Angle X-ray Diffraction (WAXD)





2.00 **Value [g liquid uptaken /g dry pulp]** 1.60 1.70 1.70 1.00 0.80 **Liquid Retention** 0.40 0.20 0.00 2% Water 50% 70% 100% 4% NaOH Ethanol

Liquid retention value Different solvents

- Liquid retention value (LRV)
 - Reflects on the liquid holding capacity

$$LRV(g/g) = \frac{m_{wet} - m_{dry}}{m_{dry}}$$

 m_{wet} represented the mass of the pulp before drying (after centrifugation) m_{drv} the dry weight of the pulp

- The best results were achieved with NaOH 2%
- At ethanol 100% is the lowest uptake

With increasing ethanol content, the LRV is decreasing





Swelling - Water

- Uniformly swollen along the length
- Width between 50-60µm
- Flexible and soft structure



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- Flexible and soft structure

Swelling - Water









Swelling – Ethanol (EtOH) 50% to 100%

- Uniformly swollen with increasing water content
- Width 30-50µm, more even distributed with increasing water content
- Dried out to soft structure (100% \rightarrow 50% EtOH)



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Swelling Ethanol 50%



Swelling Ethanol 100%







Swelling – Sodium hydroxide (NaOH) 2wt% and 4wt%

- Uniformly swollen along the length
- Width between 40-65µm
- Flexible and soft structure



Swelling NaOH 2wt%



- Uniformly swollen along the length
- Width between 40-65µm
- Flexible and soft structure

Results

Swelling NaOH 4wt%







Copyright Comparison of the LRV values 2.00 EtoH 100% and NaOH 2wt% Difference is noticeable on the surface and the morphology EtOH 100% Dired out and brittle structure NaOH 2wt% Swollen and soft structure 0.00





Swelling Ethanol 100%



Swelling NaOH 2wt%



EtOH 100%

Difference is noticeable on the

surface and the morphology

1000,0µm





Crystallinity Index Wide Angle X-ray Diffraction (WAXD)

- Investigation of crystalline and amorphous part
 - Reflects on flexibility (amorphous part) and strength (crystalline part)
- Crystallinity Index slightly increases with treatment
- Dissolution of amorphous components like hemicelluloses by NaOH and EtOH.

The reduction in amorphous content leads to a relative increase in the crystalline background.







Conclusions

What we learned...

- NaOH 2wt% resulted in the highest uptake capacity in the fibers
- Pure ethanol caused the fibers to have the lowest uptake capacity
- Compared to water and NaOH solution, ethanol and ethanol solutions swell the fibers unevenly along their length
- The slight increase in Crystallinity Index suggests that NaOH and EtOH treatments dissolve amorphous components, such as hemicelluloses, resulting in a higher relative crystalline content

Outlook

- Screening for matrix material (additives) and other swelling agents
- Testing the swelling agents with the matrix material on holocellulose fibers
- Impregnation with additives or matrix material after swelling
- Study of methods for the analysis of impregnated fibers for biocomposites











Thank you!

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[6] ISO 23714:2014 - "Pulp — Determination of water retention value (WRV)