



# Rechargeable oxygen ion batteries based on mixed conducting oxides

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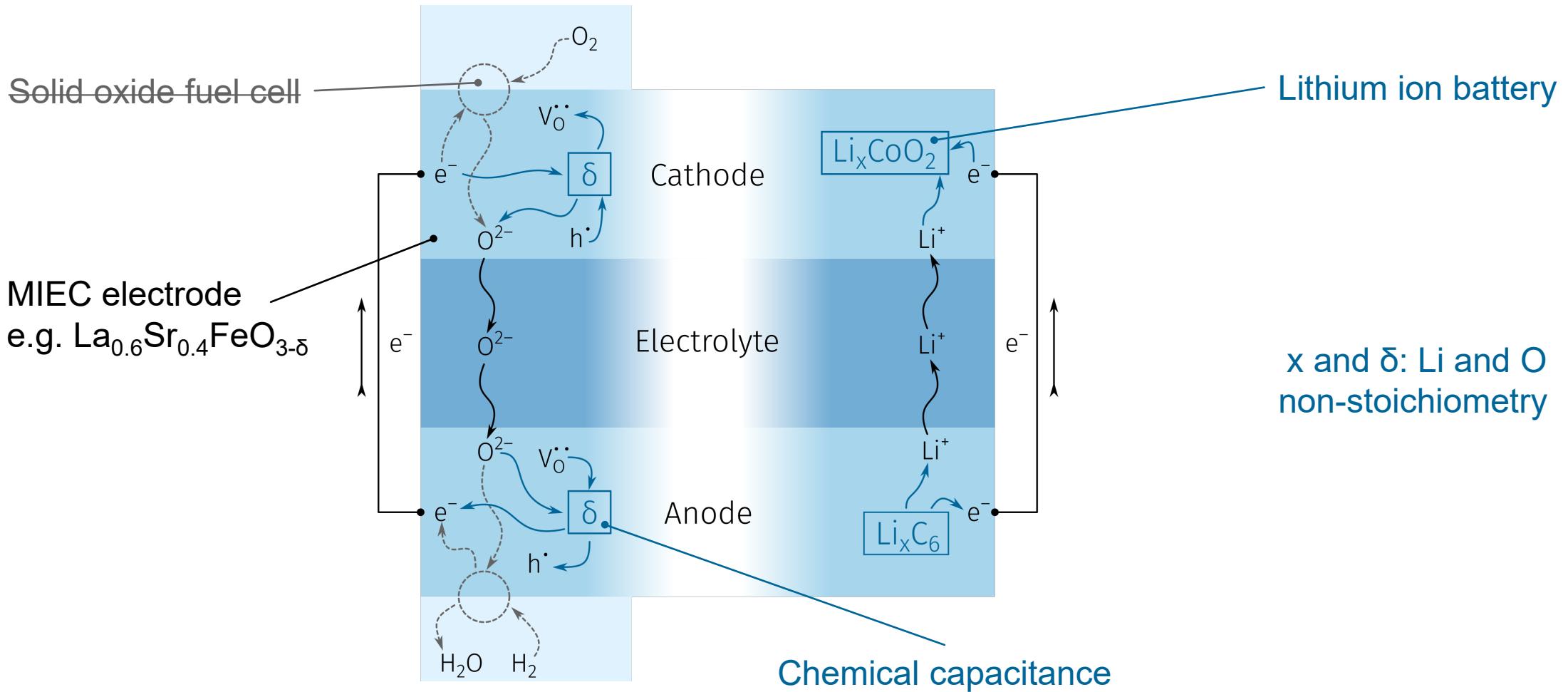
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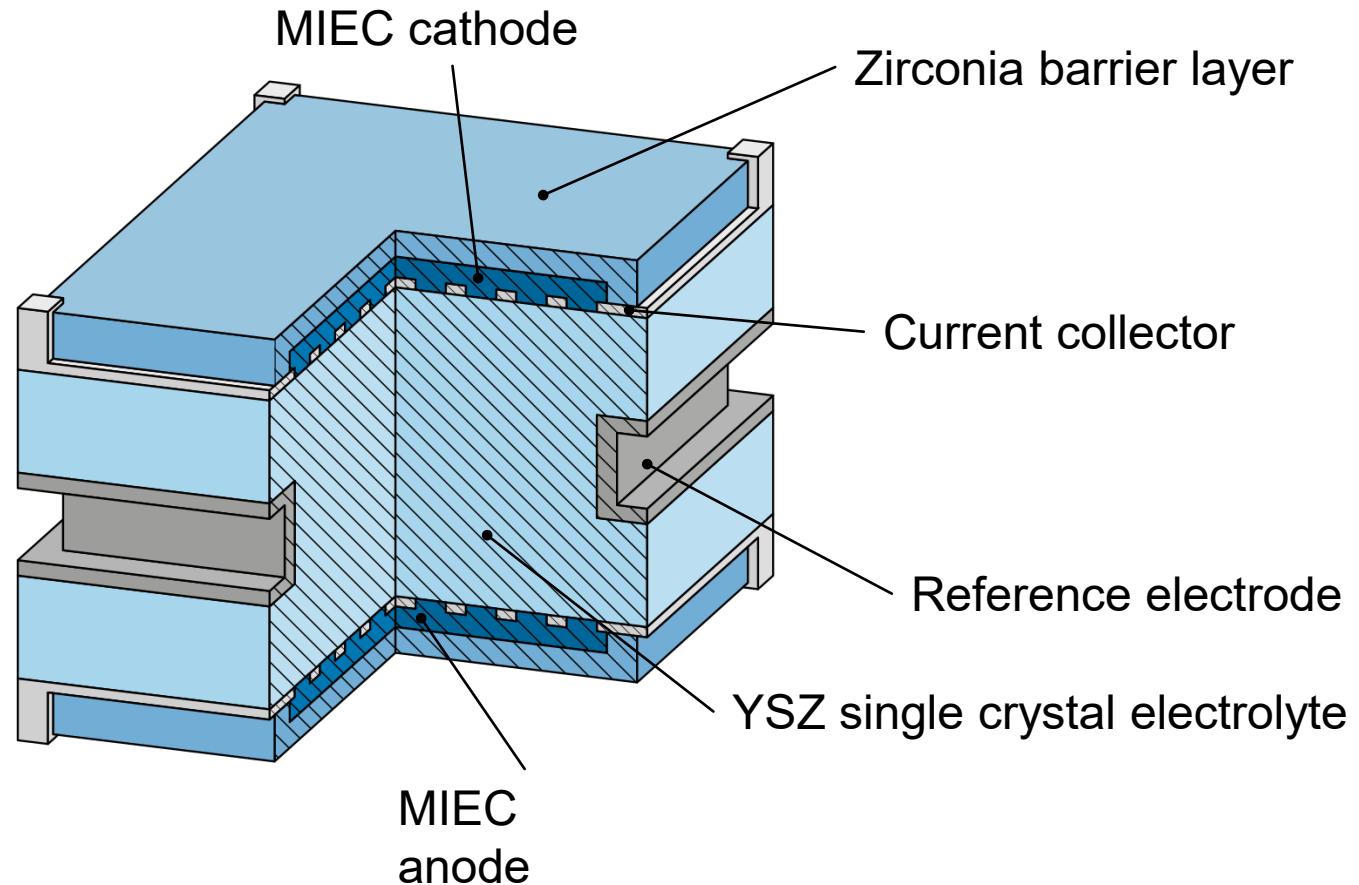


# Operation principle



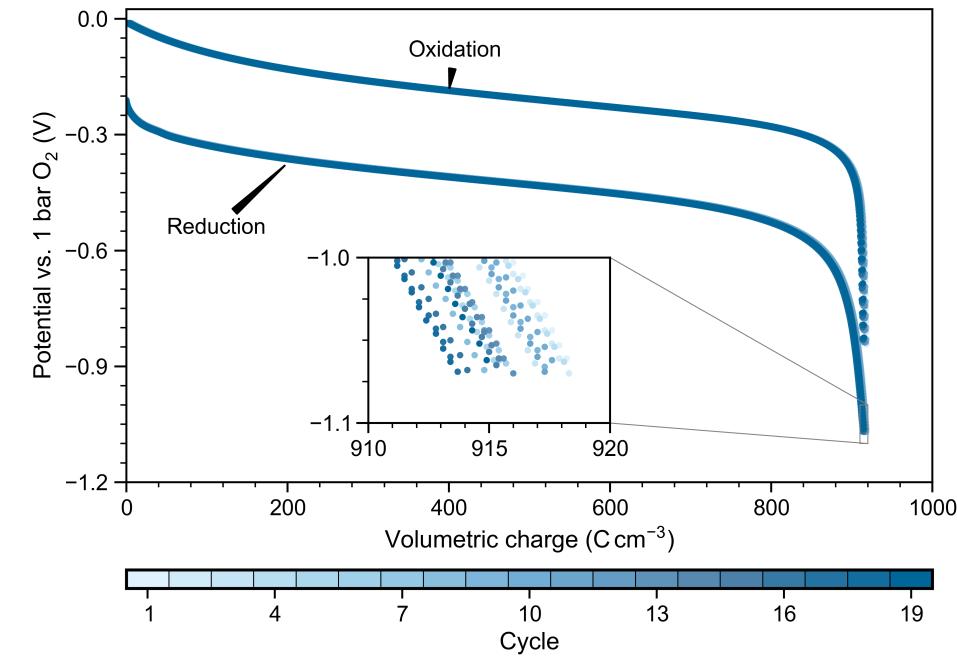
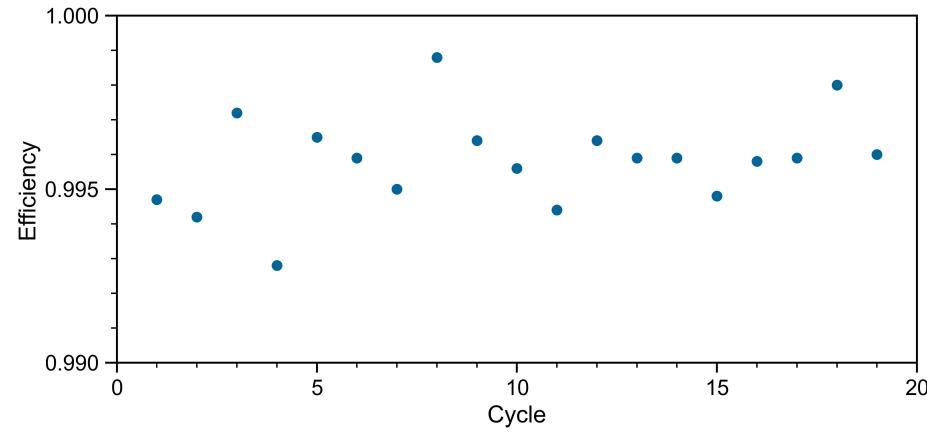
# Experimental approach

- Thin film model systems
- Pulsed laser deposition
- Three electrode setup
  
- Electrochemical impedance spectroscopy
- DC charge/voltage measurements



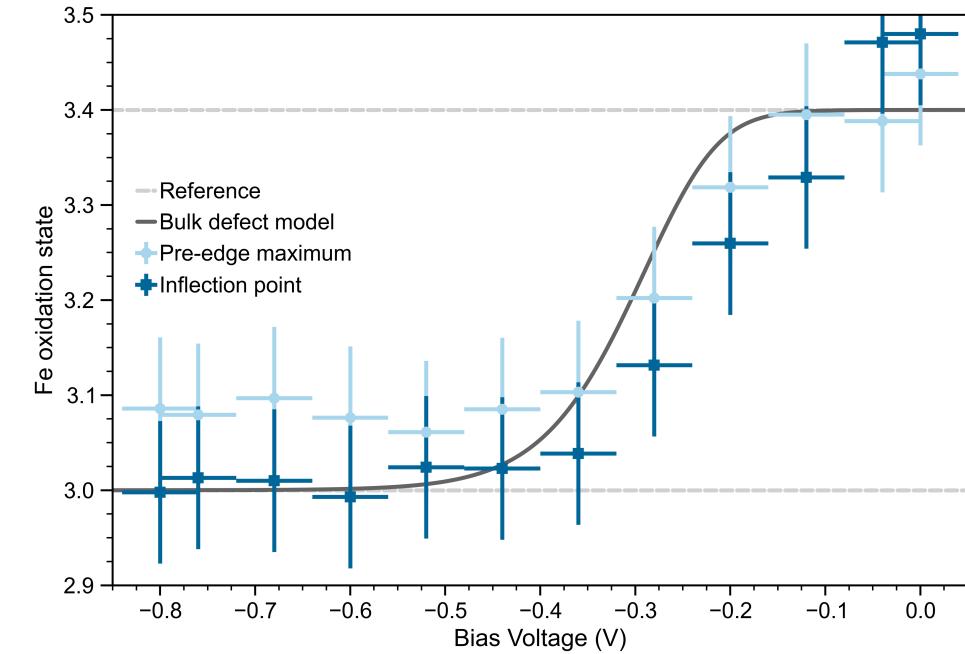
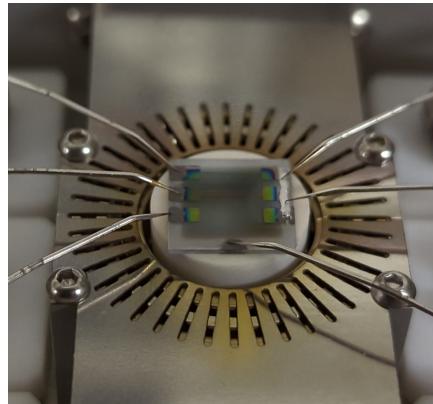
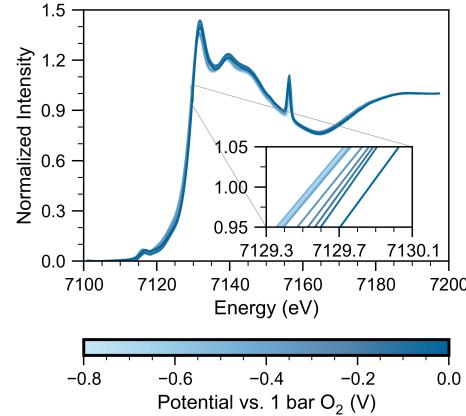
# $\text{La}_{0.6}\text{Sr}_{0.4}\text{FeO}_{3-\delta}$ electrodes - Charge voltage characteristics

- 915  $\text{C cm}^{-3}$  at -0.3 V vs. 1 bar  $\text{O}_2$  ( $255 \text{ mA h cm}^{-3}$ )
- <0.1 % capacity loss per cycle
- >99 % coulomb efficiency



450 nm LSF film, 350 °C, 58  $\mu\text{A cm}^{-2}$  (5 C), 25 Pa  $\text{O}_2$

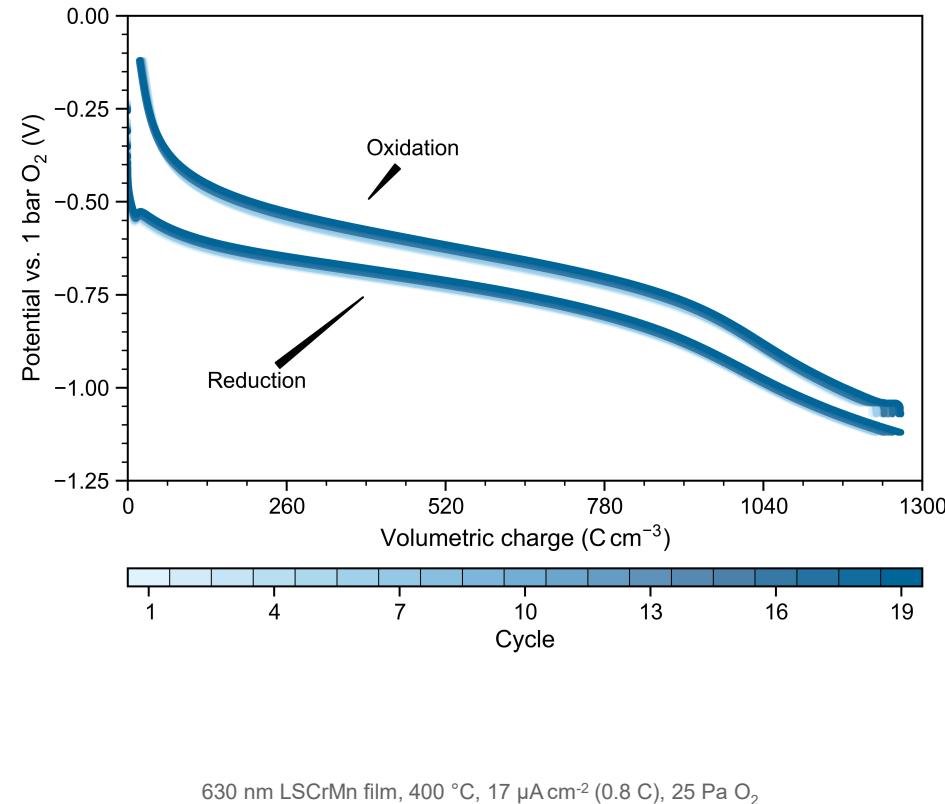
# In-situ synchrotron X-ray absorption spectroscopy



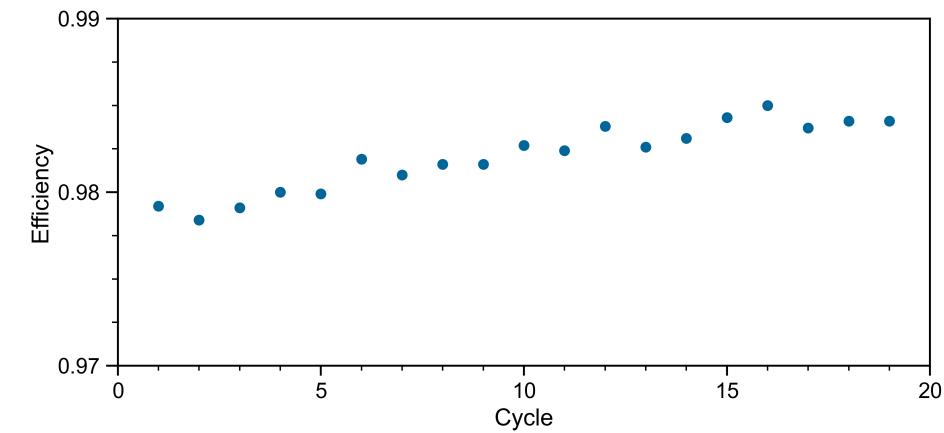
200 nm LSF film, 350 °C, 100 kPa O<sub>2</sub>

- Electrode charge/discharge leads to Fe oxidation state change
- $2 \text{Fe}_\text{Fe}^x + \text{V}_\text{O}^{\cdot\cdot} + \text{O}_\text{O}^{2-} \rightleftharpoons 2 \text{Fe}_\text{Fe}^{\cdot} + \text{O}_\text{O}^x + 2\text{e}^-$
- Agrees with bulk defect model

# $\text{La}_{0.5}\text{Sr}_{0.5}\text{Cr}_{0.2}\text{Mn}_{0.8}\text{O}_{3-\delta}$ anodes

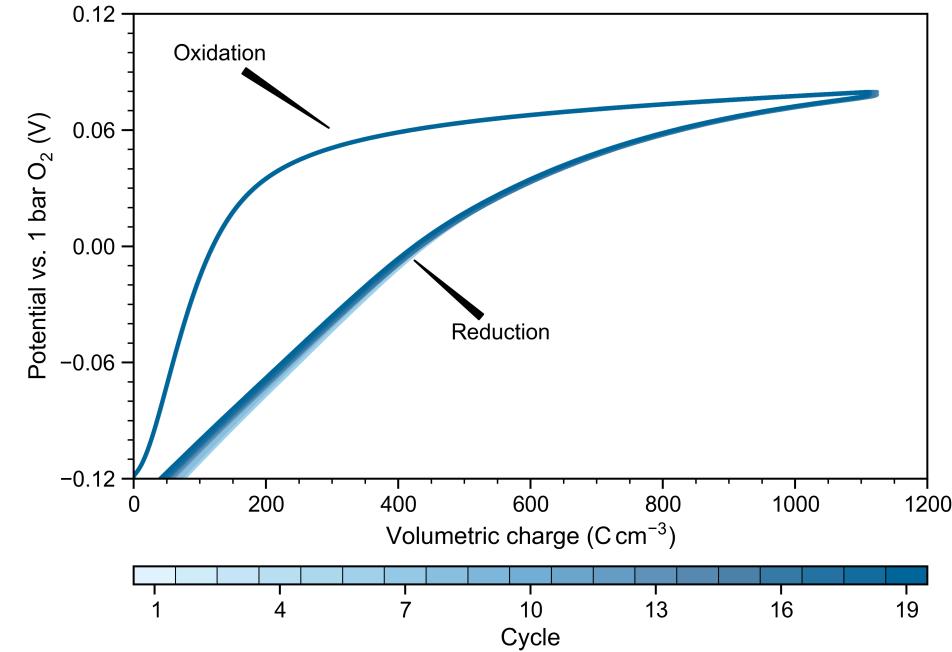
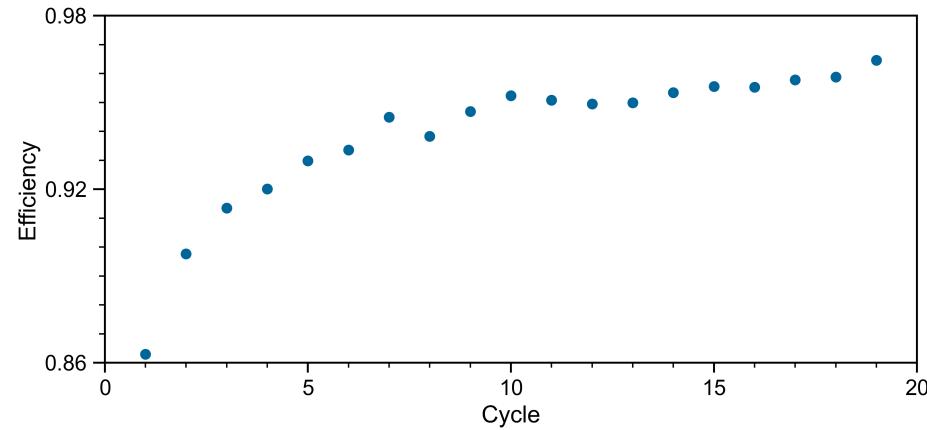


- 1250  $\text{C cm}^{-3}$  at -0.6 V vs. 1 bar  $\text{O}_2$  (350  $\text{mA h cm}^{-3}$ )
- Up to 3500  $\text{C cm}^{-3}$  at -2 V
- Poster by B. Wagner



# La<sub>2</sub>NiO<sub>4+δ</sub> cathodes

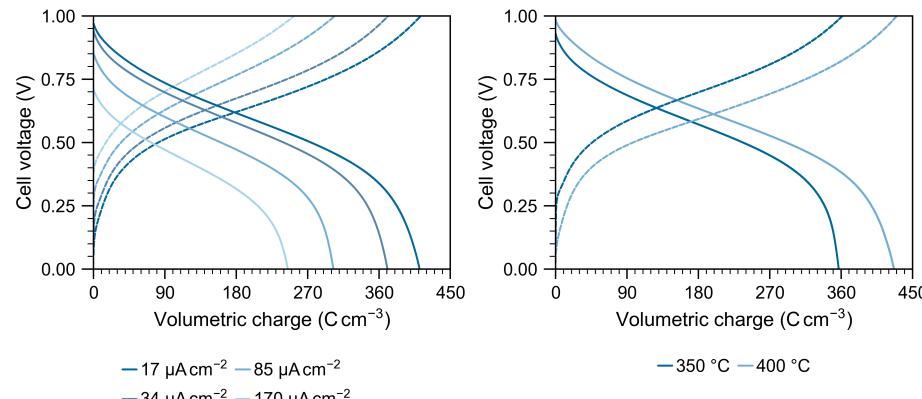
- Intercalation of oxygen as interstitials
- High half cell potential
- $2 \text{ Ni}_{\text{Ni}}^{\text{x}} + \text{O}^{2-} \rightleftharpoons 2 \text{ Ni}_{\text{Ni}}^{\cdot} + \text{O}_{\text{i}}^{\prime\prime} + 2\text{e}^{-}$



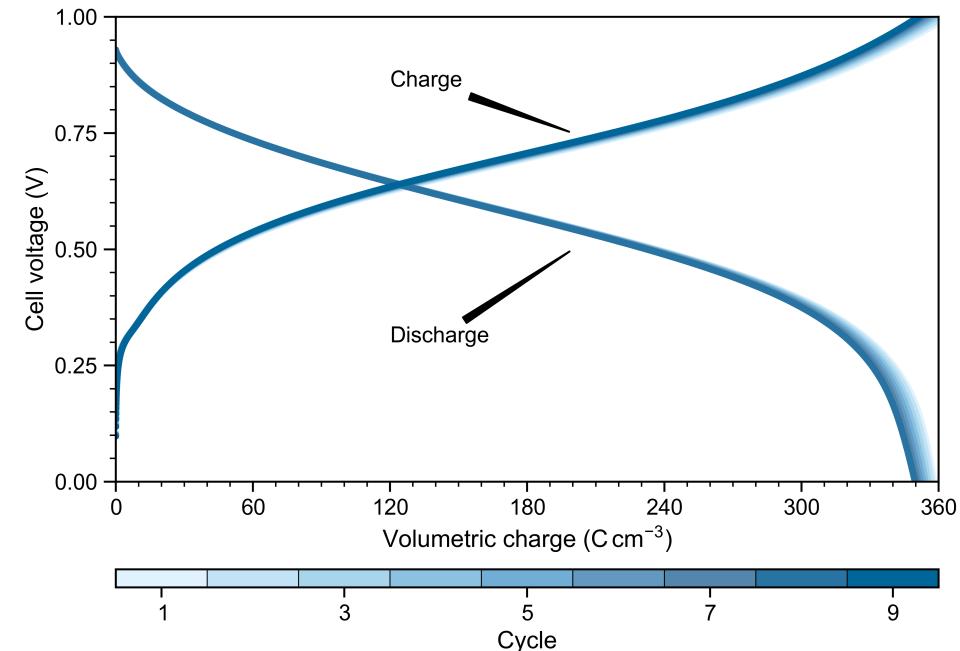
200 nm LNO film, 400 °C, 17 μA cm<sup>-2</sup>, 1000 Pa O<sub>2</sub>, prepared by LMGP, Grenoble

# Complete oxygen ion battery: LScrMn | YSZ-SC | LSF

- 420 C cm<sup>-3</sup> at 0.6 V
- 250 J cm<sup>-3</sup>
- >99 % coulomb efficiency



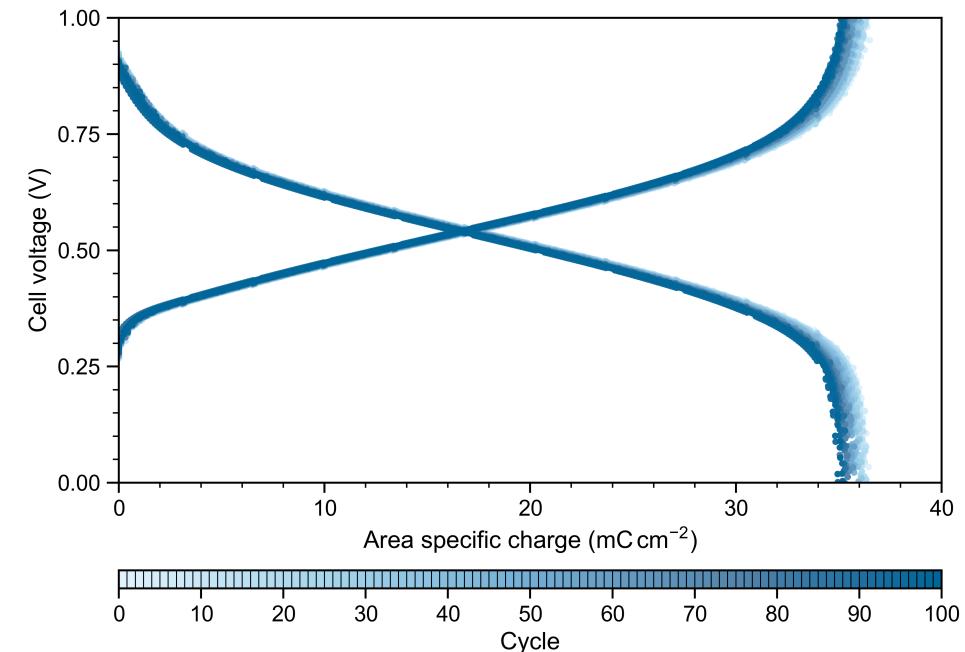
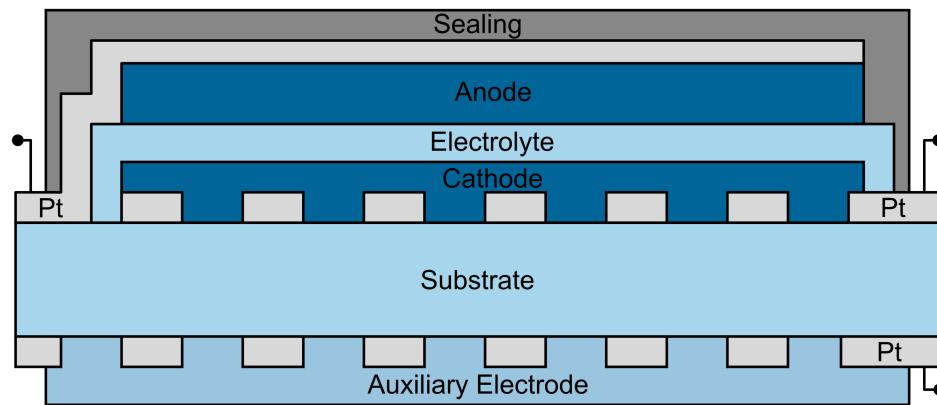
320 nm LSF, 192 nm LScrMn, 25 Pa O<sub>2</sub>



320 nm LSF, 192 nm LScrMn, 350 °C, 8.5 μA cm<sup>-2</sup> (1.6 C), 25 Pa O<sub>2</sub>

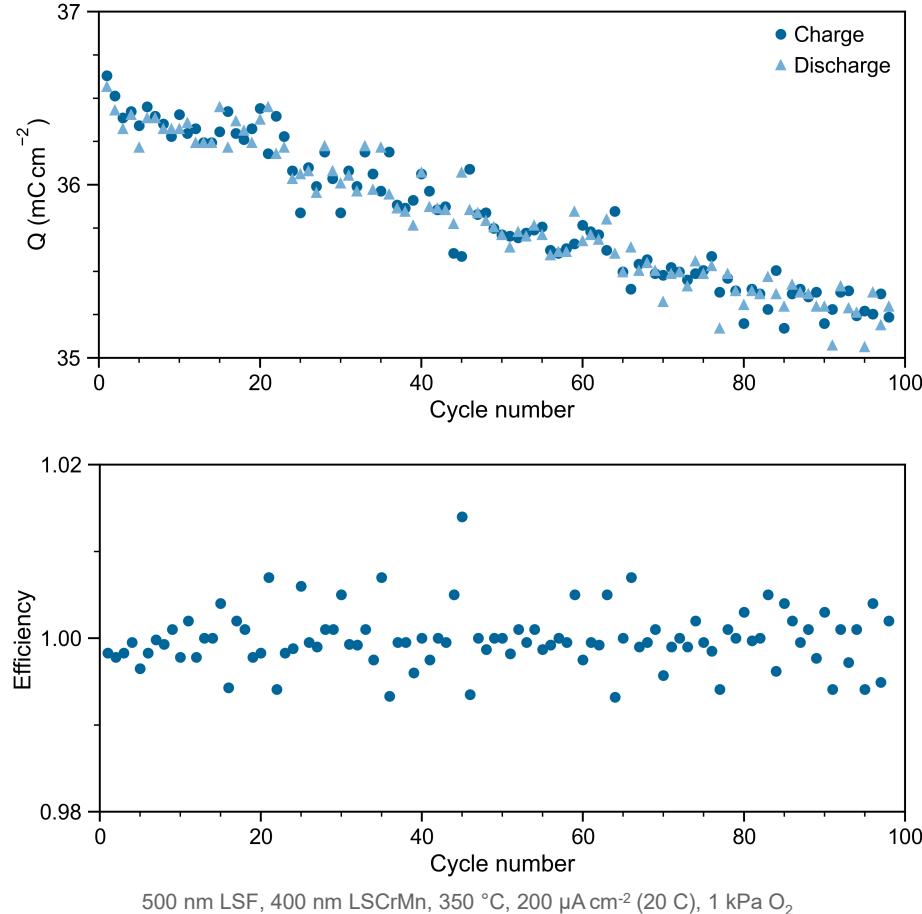
# Full cell battery with thin film electrolyte

- Thin film electrolyte – 800 nm
- Reactive DC sputtering from Y/Zr metal target

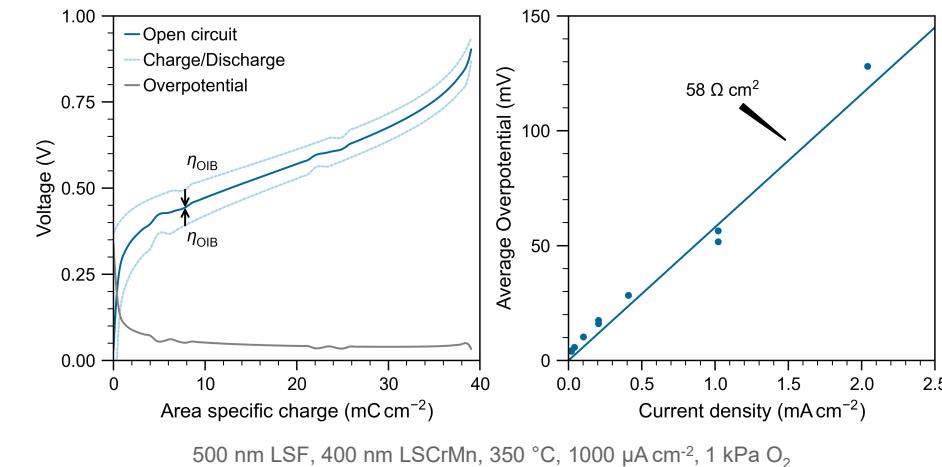


500 nm LSF, 400 nm LScrMn, 350 °C, 200  $\mu\text{A cm}^{-2}$  (20 C), 1 kPa O<sub>2</sub>

# Full cell battery with thin film electrolyte

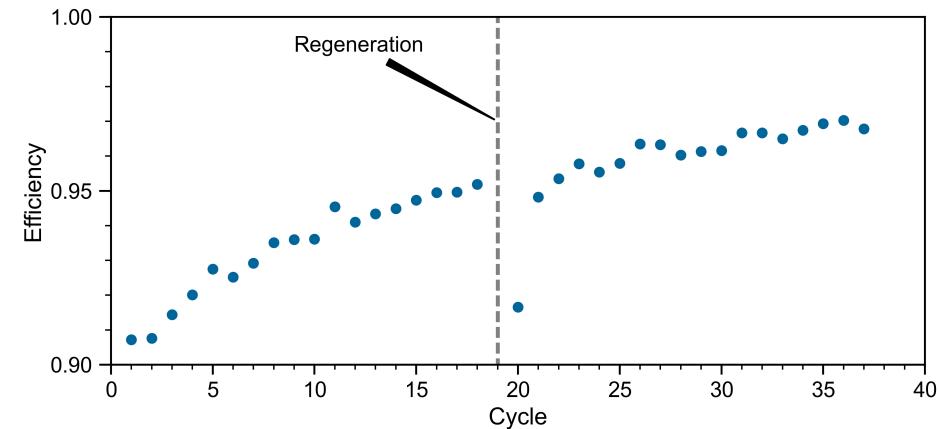
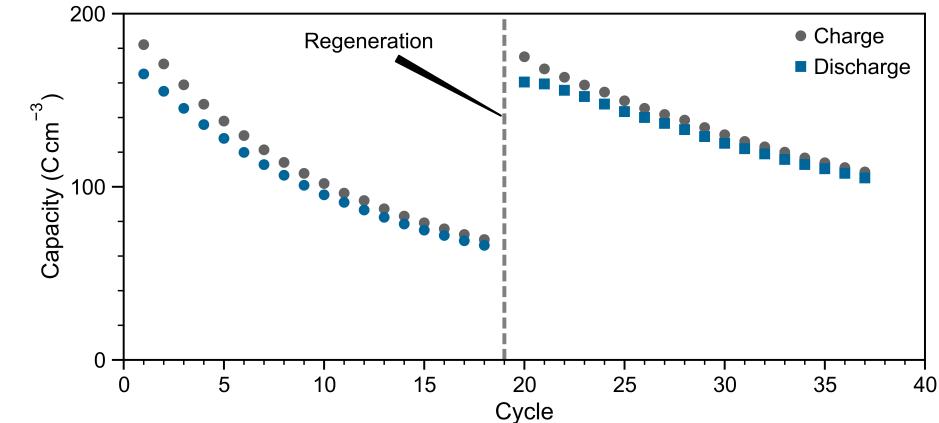
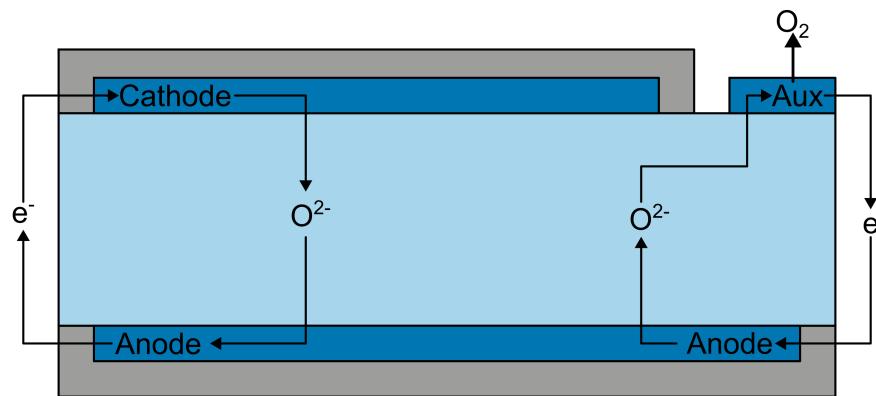


- Interface resistance relevant
- < 0.05 % charge lost per cycle
- Cell stable over >2000 cycles in total (with regeneration steps)



# Regeneration of degraded cell capacity

- Removal of leaked oxygen via auxiliary electrode
- Cell capacity repeatedly regenerable

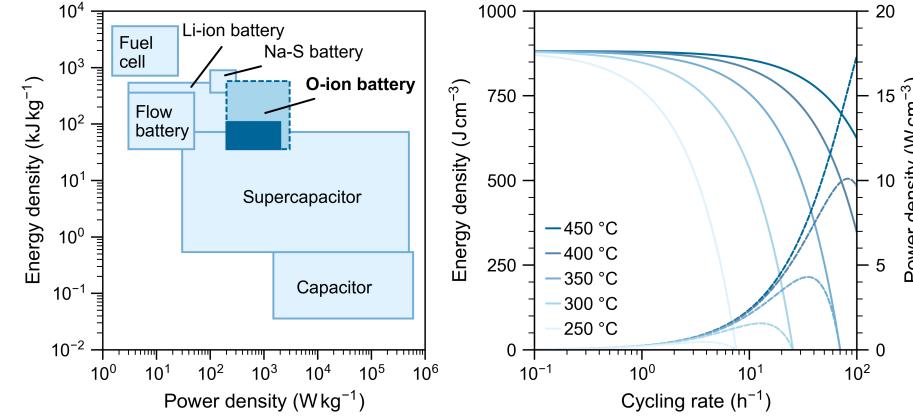


320 nm LSF, 192 nm LSCrMn, 350 °C, 17  $\mu\text{A cm}^{-2}$  (3.2 C), 1 kPa  $O_2$

# The potential of oxygen ion batteries

- Cell voltage: 1V
- Electrode capacity:  $2250 \text{ C cm}^{-3}$
- $1 \mu\text{m}$  electrolyte
- $25 \mu\text{m}$  electrodes

- Energy:  $900 \text{ J cm}^{-3}$
- Charge rate:  $1 - 100 \text{ h}^{-1}$
- Power:  $1 - 18 \text{ W cm}^{-3}$



- Abundant elements (Fe, Cr, Mn, Ti, Ca, Sr, ...)
- Non-flammable, non-toxic oxides
- Regenerable cell capacity

# Conclusions

- High temperature energy storage via variable oxygen stoichiometry
- Non-flammable, non-toxic oxides, not reliant on critical raw elements
- Cell capacity repeatedly regenerable via atmosphere exchange

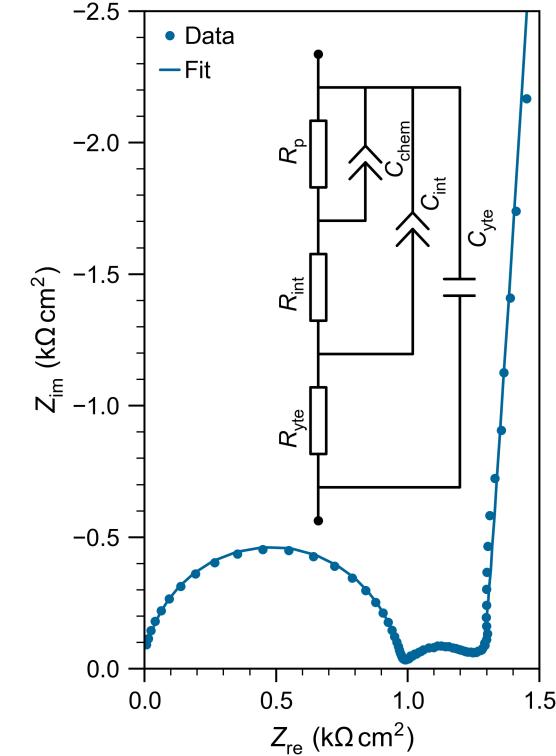
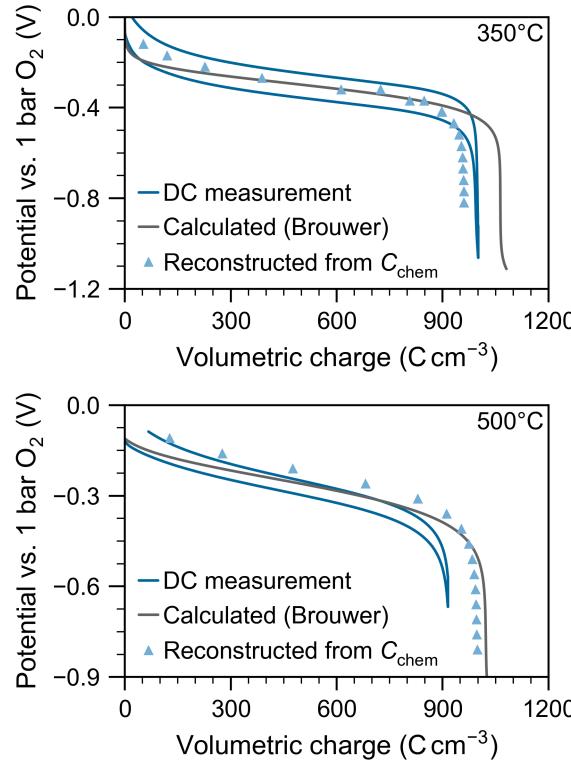
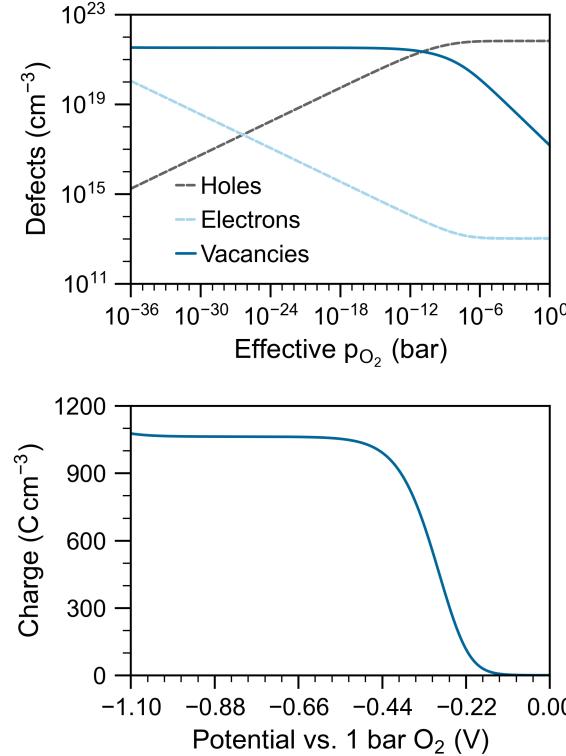
# Thank you for your attention!

For more information, please visit :

- Schmid, M. Krammer, and J. Fleig, “Rechargeable Oxide Ion Batteries Based on Mixed Conducting Oxide Electrodes,” *Advanced Energy Materials*, **2023**, 13, 2203789.



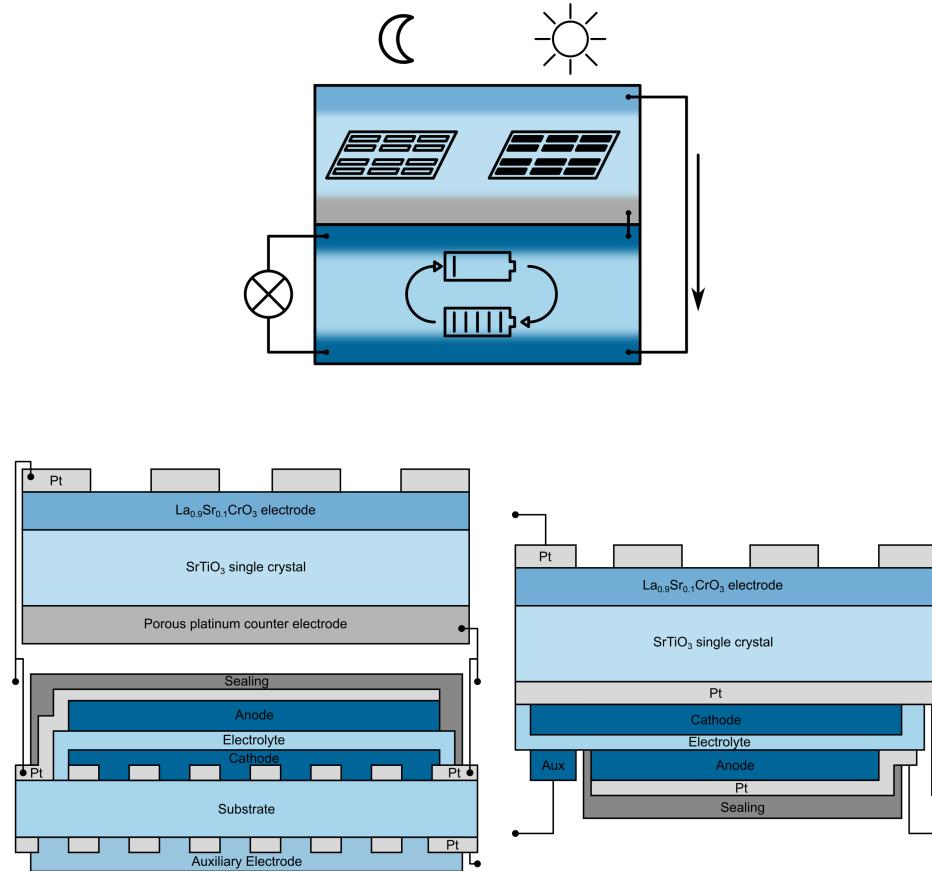
# LSF electrodes – A defect chemical explanation



$$\Delta Q = 2F[V_O^\cdot] \quad U = \frac{RT}{4F} \ln \left( \frac{p_{\text{O}_2}}{1 \text{ bar}} \right)$$

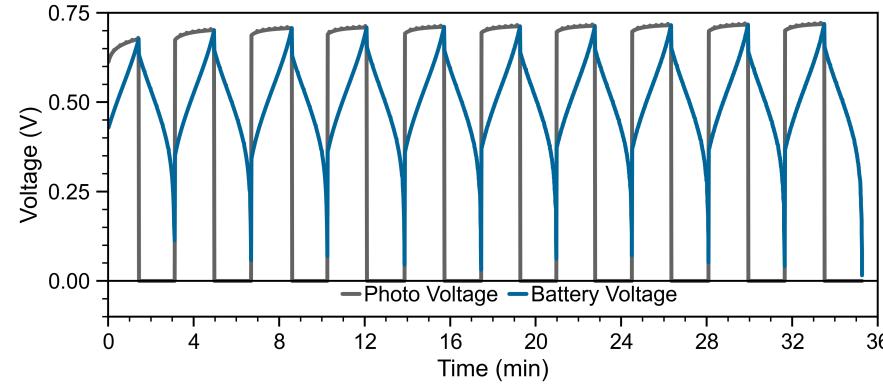
$$Q(U) = \int C_{\text{chem}} dU \quad C_{\text{chem}} = \frac{F^2}{RT} \left( \frac{1}{4[V_O^\cdot]} + \frac{1}{[h^\cdot]} \right)^{-1}$$

# The Harvestore concept



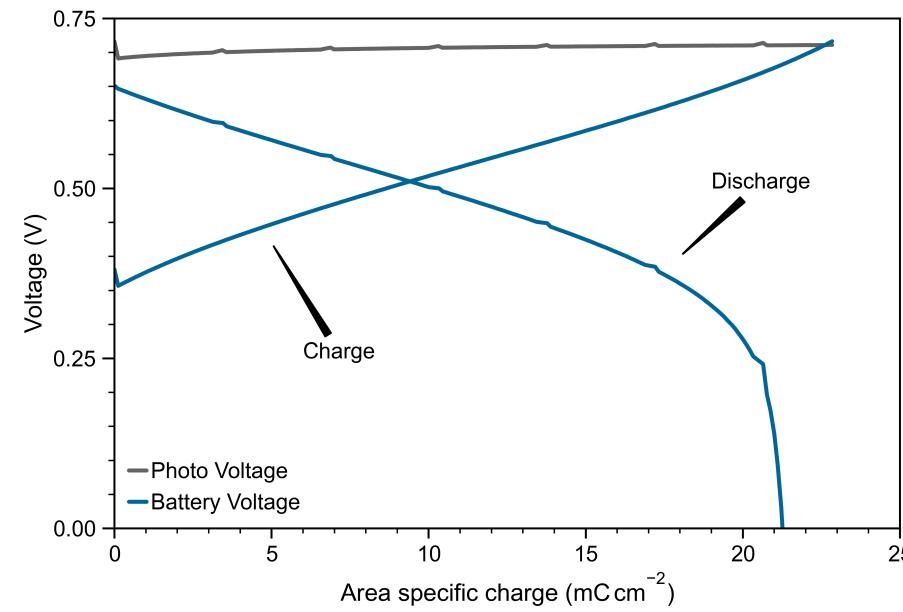
- Thin film oxygen ion battery coupled to SrTiO<sub>3</sub> PV cell
- Battery charged by photo-current under illumination
- Discharged without illumination

# Combined power harvesting and storage



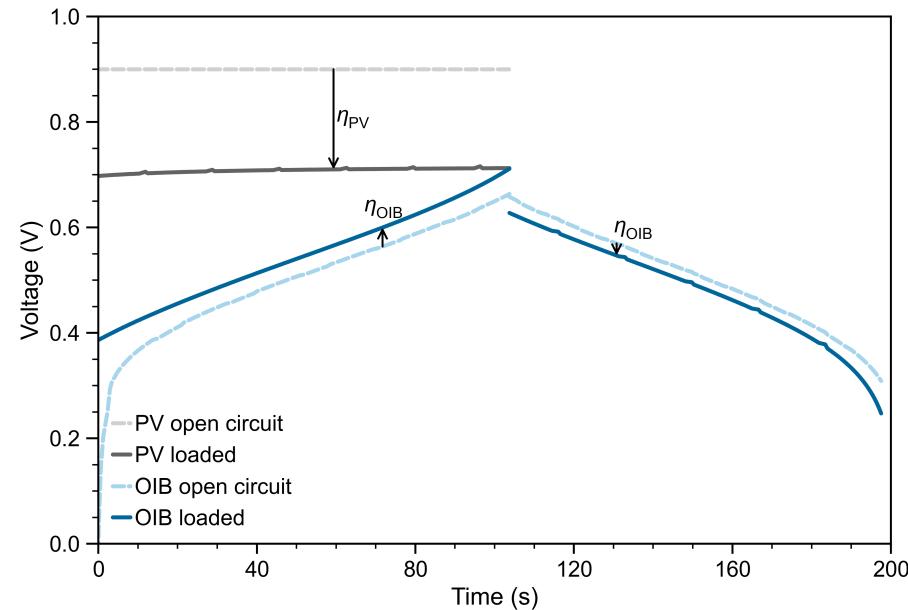
- OIB repeatedly charged with PV
- Discharged without illumination

500 nm LSF, 400 nm LScrMn, 350 °C, 200  $\mu\text{A cm}^{-2}$  (36 C), 25 Pa O<sub>2</sub>, separate cells



500 nm LSF, 400 nm LScrMn, 350 °C, 200  $\mu\text{A cm}^{-2}$  (36 C), 25 Pa O<sub>2</sub>

# Harvestore characteristics



- Up to  $200 \mu\text{A cm}^{-2}$  (36 C)
- Up to  $10 \text{ mJ cm}^{-2}$
- 86 % energy storage efficiency

500 nm LSF, 400 nm LScrMn, 350 °C, 200  $\mu\text{A/cm}^2$  (20 C), 25 Pa  $\text{O}_2$ , separate cells