

STREAMLINED CO₂ CONVERSION FOR MODERN CARBONYLATION CHEMISTRY

Giada Moroni^{1*}; Kristof Stagel¹ and Katharina Bica-Schröder¹

¹ Institute of Applied Synthetic Chemistry, TU Wien, Getreidemarkt 9, 1060 Vienna, Austria

*Presenting author: giada.moroni@tuwien.ac.at

Carbon monoxide (CO) is widely employed in catalytic carbonylative reactions to efficiently introduce carbonyl groups into a variety of molecules, generating a plethora of high-value fine chemicals. However, its toxicity and handling risks pose significant challenges. [1] In recent years, carbon dioxide (CO₂) has emerged as a promising, non-toxic, and abundant alternative for CO-based transformations. [2] Moreover, flow technology has revolutionised reactions involving toxic gases and hazardous chemicals, providing improved efficiency, enhanced safety, and precise reaction parameters control. [3]

Here, we present a continuous flow chemistry platform for carbonylation reactions, in which CO is *in situ* generated using the photo- and electrochemical reduction of CO₂. [4] A series of carbonylation processes were investigated under continuous flow conditions (**Figure 1**). As a result of a comprehensive assessment of parameters, we identified optimal reaction conditions that enabled rapid transformations, high selectivity, and good to excellent yields. Our findings highlight the potential of CO₂ valorisation in carbonylation chemistry and demonstrate the synergy with continuous flow technology, paving the way for safer and more sustainable carbonylation methodologies.

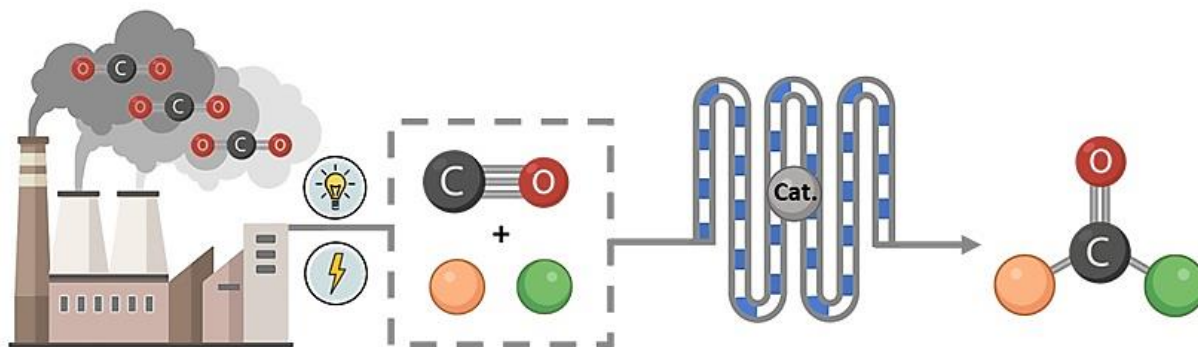


Figure 1. CO₂ utilisation and continuous flow platform for carbonylation reactions.

References

- [1] Peng, J-B., Geng, H-Q., Wu, X. F., *Cell J.*, 5, 526 (2019)
- [2] Liu, Q., Wu, L., Jackstell, R., Beller, M., *Nat. Commun.*, 6, 5933, (2015)
- [3] Mallia, C. J., Baxendale, I. R., *OPRD*, 20, 327 (2016)
- [4] Stagel, K., Rath, K., Kathe, P. M., Schnürch, M., Huber, T. M., Opitz, A. K., Bica-Schröder, K., *Angew. Chem. Int. Ed.*, e202420578 (2025)